Access to H.E. National Programme Unit



Unit Title:	Science: Introduction to Chemistry	Ungraded Unit Code:	UD23DEV08
Pathway(s):	All Pathways		
Module(s):	Developmental		
Level:	2	Credit Value:	3
Valid from:	11 th November 2019	Valid to:	31 st July 2025

LEARNING OUTCOMES	ASSESSMENT CRITERIA	
The learner will:	The learner can:	
Understand atomic structure	1.1 Describe the structure of an atom	
	Explain the differences between protons, electrons and neutrons in terms of there relative mass and electric charge	
	Identify elements and isotopes from atomic and mass numbers	
	1.4 Explain the difference between an element, a compound and a mixture	
2.2 2.3	2.1 Present the electronic structures of any of the first twenty elements and relate outer electrons to group number	
	Identify metals and non-metals in the s-block and p-block elements	
	2.3 Explain the formation of ionic bonds between a metal and a non-metal	
	2.4 Explain the formation of covalent bonds between non-metal elements	
	2.5 Describe the type of bonding found in metals	
	2.6 Explain the properties of metals	

Access to H.E. National Programme Unit



	Understand chemical reactions and balanced chemical equations	 3.1 Describe the chemical processes: a) direct combination b) combustion c) thermal decomposition d) neutralisation
		3.2 Demonstrate how to balance chemical equations
		3.3 Explain the principle of conservation of mass
		3.4 Identify the mass of a reactant or product from information about the masses of other reactants or products
4.	Understand the factors affecting the rate of a reaction	4.1 Explain the effect of temperature, concentration and surface area of a solid on the rate of a reaction
		4.2 Interpret the results of experiments to investigate the effects of temperature, concentration and surface area of a solid on the rate of a reaction
		4.3 Describe the properties of catalysts, including enzymes
		4.4 Describe three practical applications of catalysts